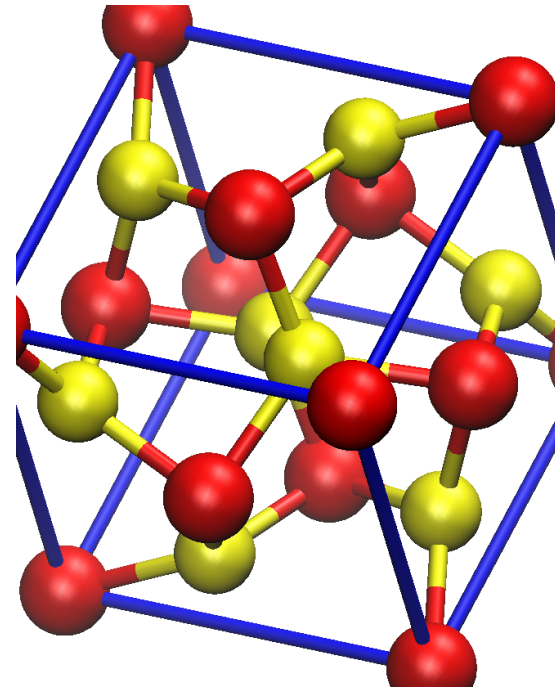
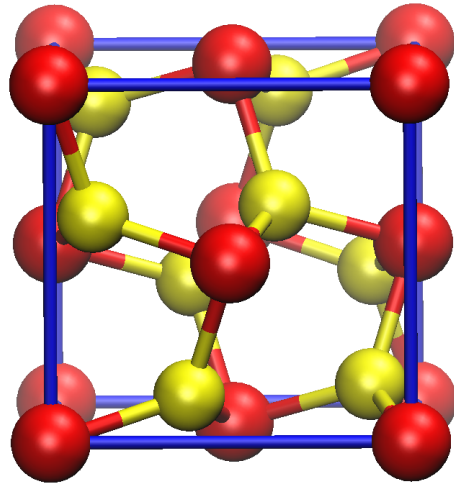


The bandgap of pyrite: bandgap oscillations on an ultrafast timescale

Brian Kolb
Alexie M. Kolpak



Pyrite (FeS_2)

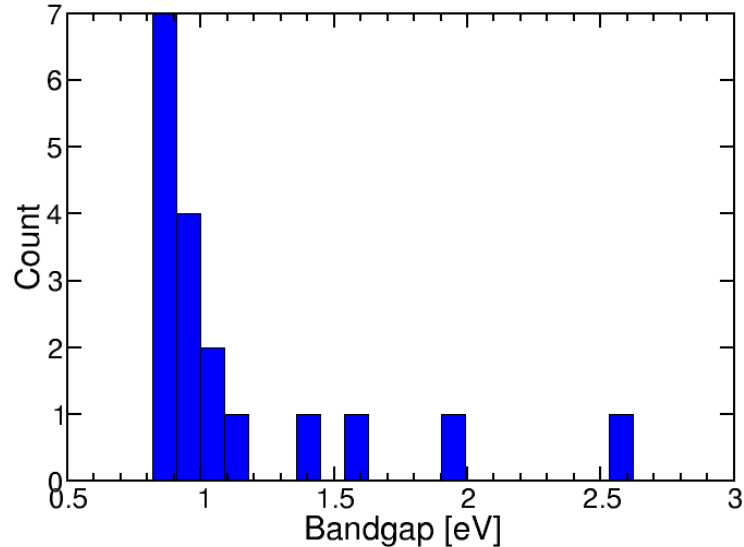


S_2 dimers along $\langle 111 \rangle$



Pyrite as a semiconductor

Histogram of experimentally measured bandgaps



Measurement Techniques

- Photocurrent
- Quantum efficiency
- Reflectance
- Optical absorption
- Transmittance

Ferrer *et al.*, Solid State Comm., **74**, 913 (1990)



Reasons to like pyrite as a solar cell material

Good bandgap (0.95 eV)

- Good at absorbing solar radiation

High absorption

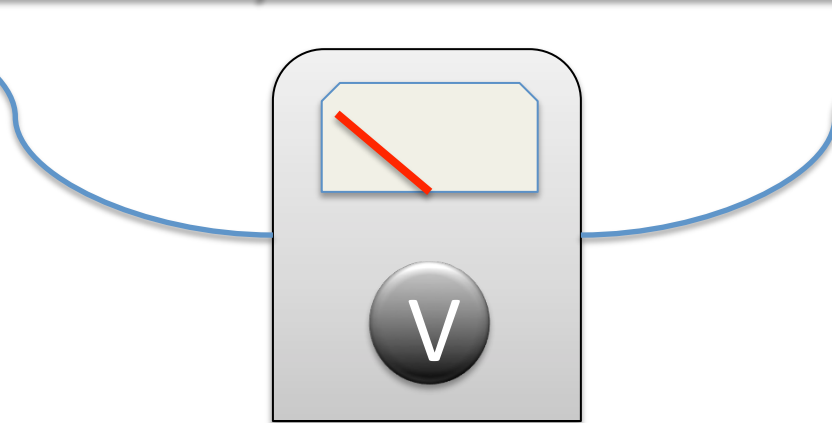
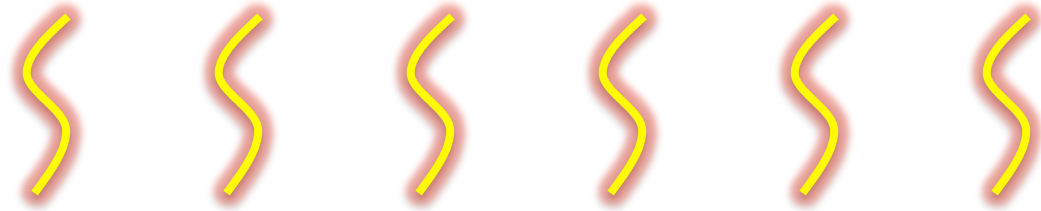
- Thin films can absorb virtually all incident radiation

Economics

- Cheap
- Non-toxic
- Abundant



Pyrite fails to perform



$$V_{OC} = 0.2 \text{ eV}$$



Possible explanations

Bulk defects

- Should be minimal

Intrinsic surface states

- Don't seem to affect bandgap

Surface defects

- Jury is still out



What we're interested in

Low open circuit voltage



Dynamical properties of the bandstructure

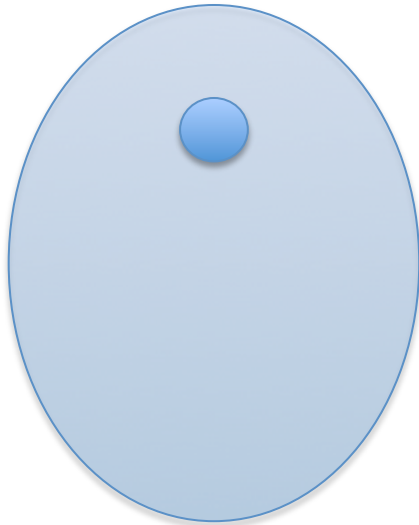


The GW method

Single particle Green's function

Electronic self-energy

$$\Sigma = iGW$$

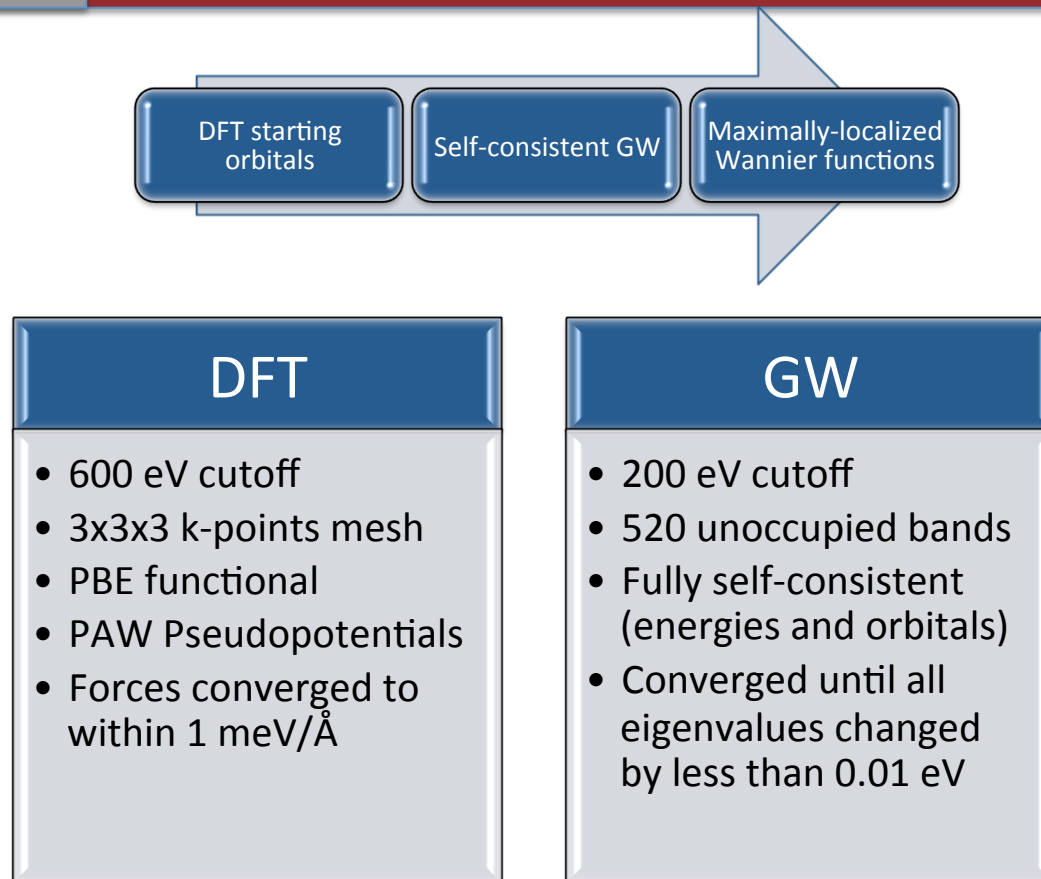


Screened Coulomb interaction

Electron plus polarization cloud →
weakly interacting quasiparticles

Expansion in screened Coulomb interaction





Accepted bandgap = 0.95 eV

GW calculations

Experimental structure: 1.01 eV

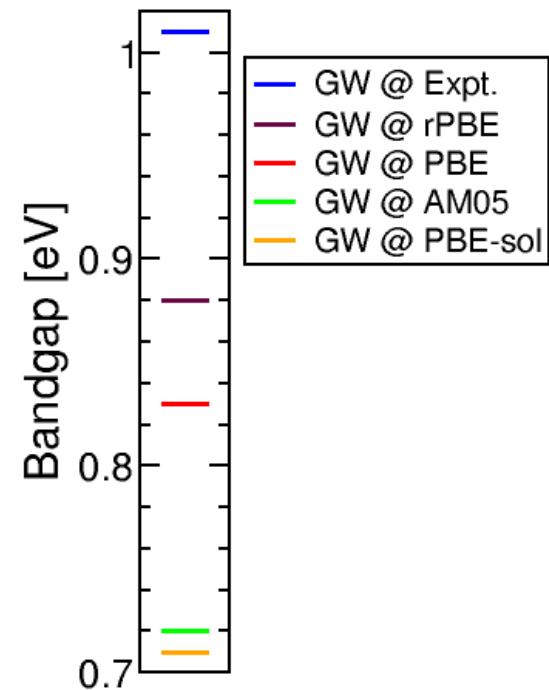
PBE-relaxed structure = 0.83 eV



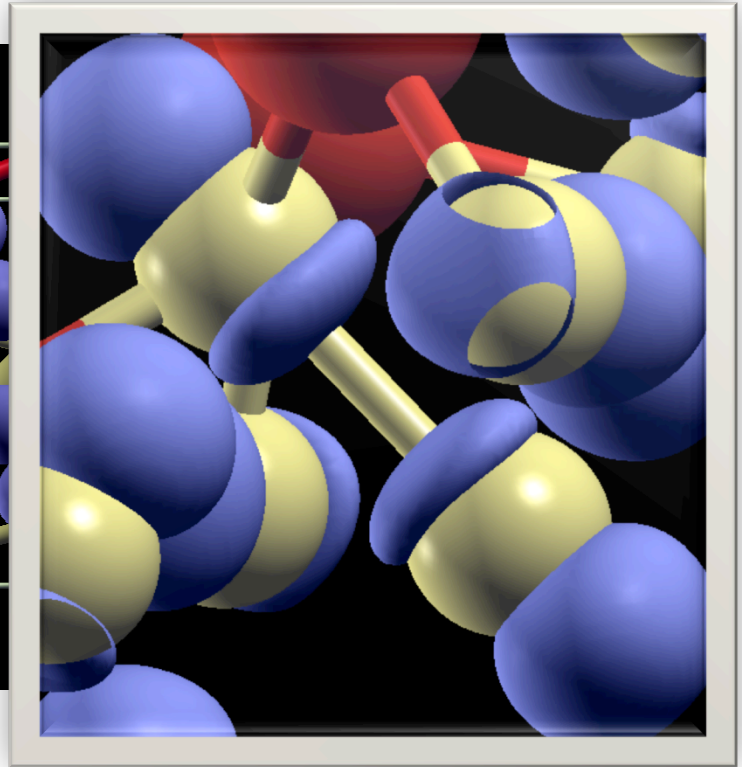
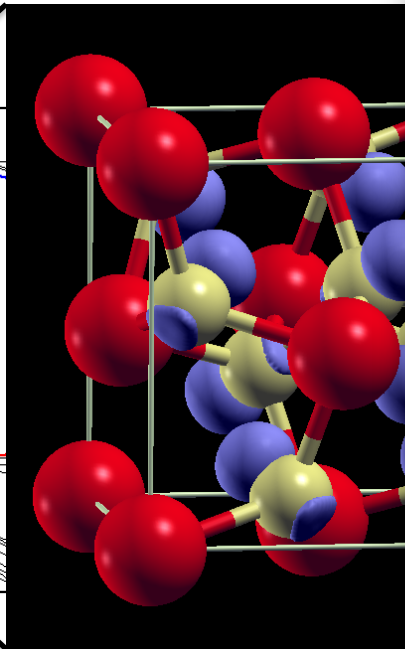
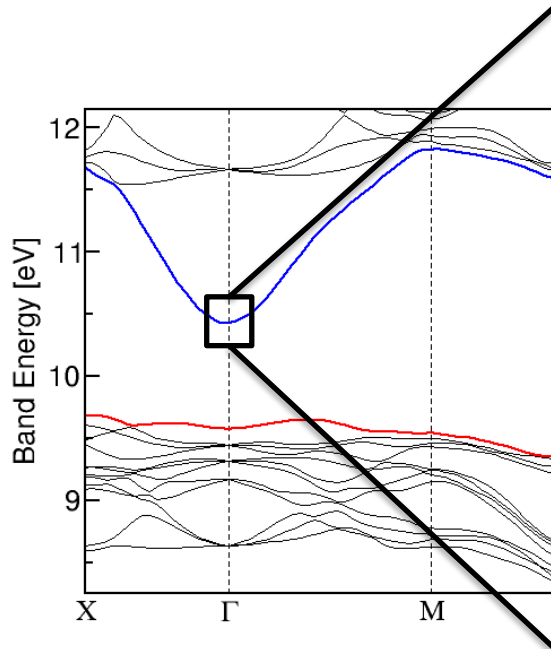
Bulk bandgap

Accepted bandgap = 0.95 eV

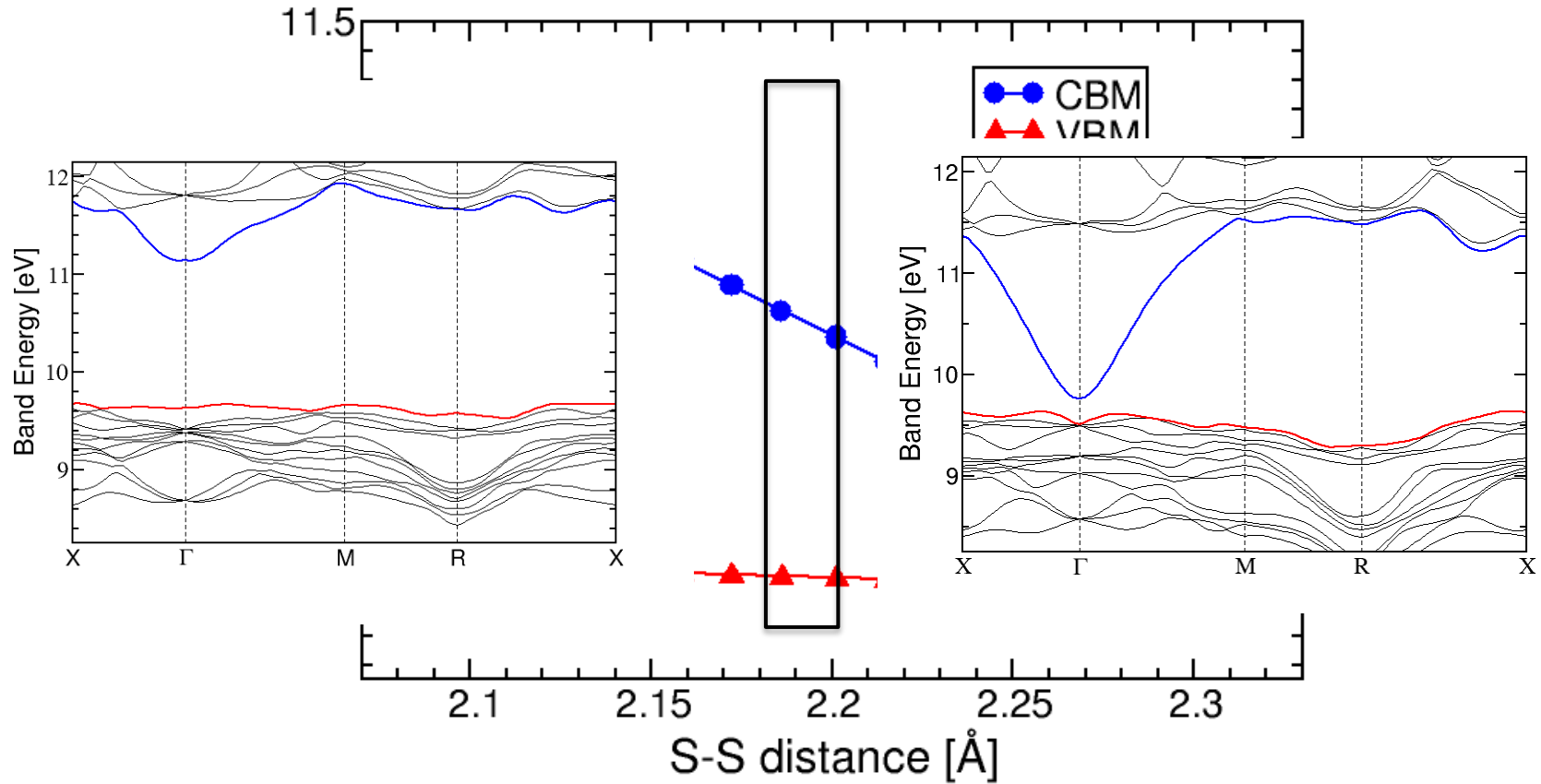
GW bandgap at different geometries



Conduction band minimum



S-S distance dependence of the gap



Why is this important?

- Bandgap is very sensitive to what coordinates are used

- Phonons

$$\omega = 347 \text{ cm}^{-1}$$

$$T = 100 \text{ fs}$$



Bulk bandgap with phonons

Thermal excitation at room temperature

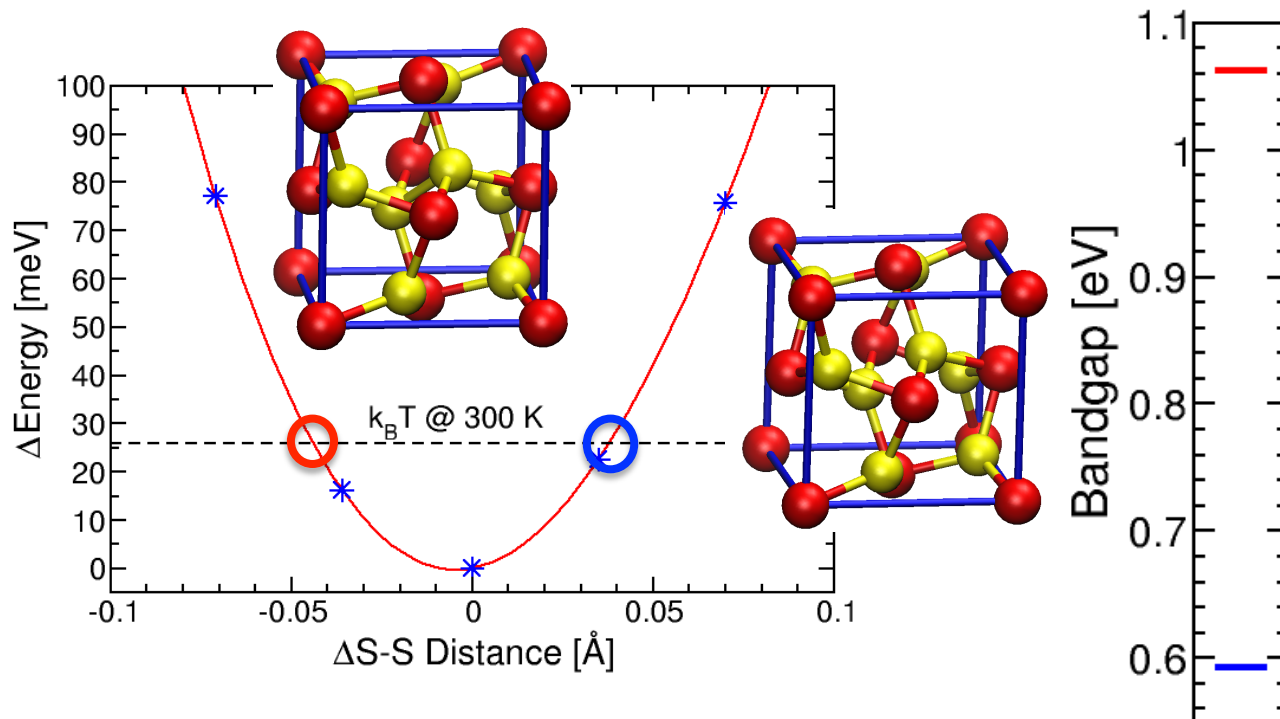
$$E_{\omega} = \hbar\omega \left(N_{\omega} + \frac{1}{2} \right)$$
$$N_{\omega} = \frac{1}{\left(e^{\frac{\hbar\omega}{k_B T}} - 1 \right)}$$

$$\Delta E = 26 \text{ meV} \approx k_B T$$

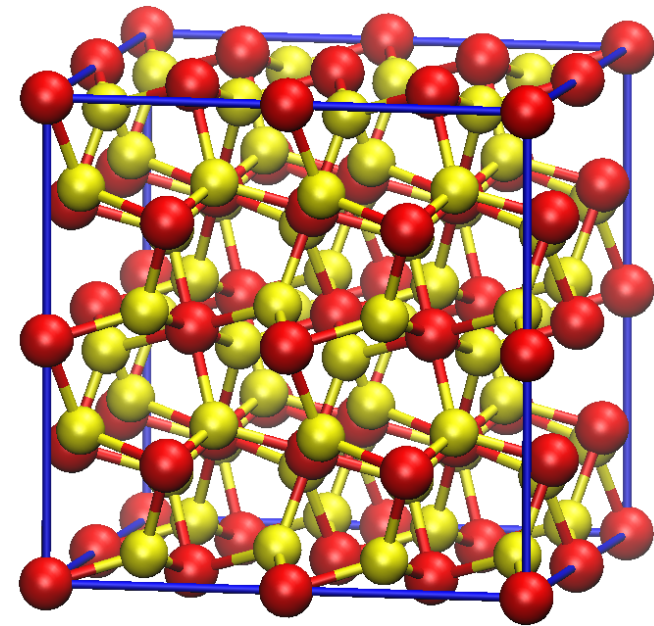
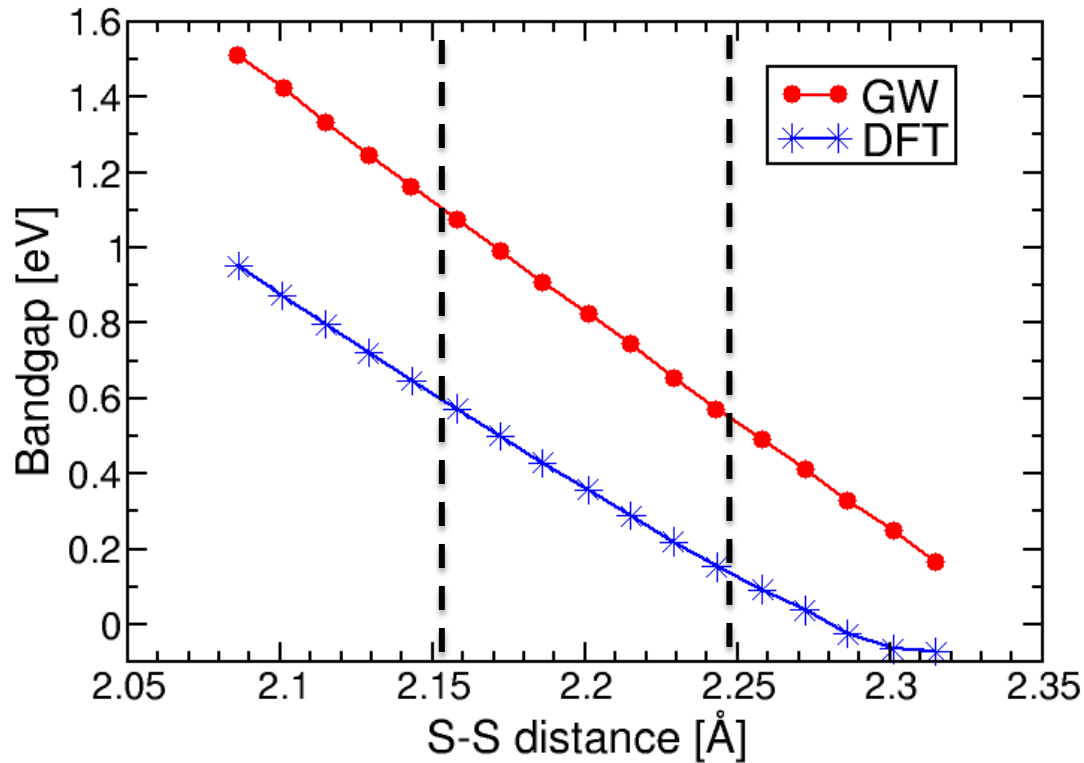


Bulk bandgap with phonons

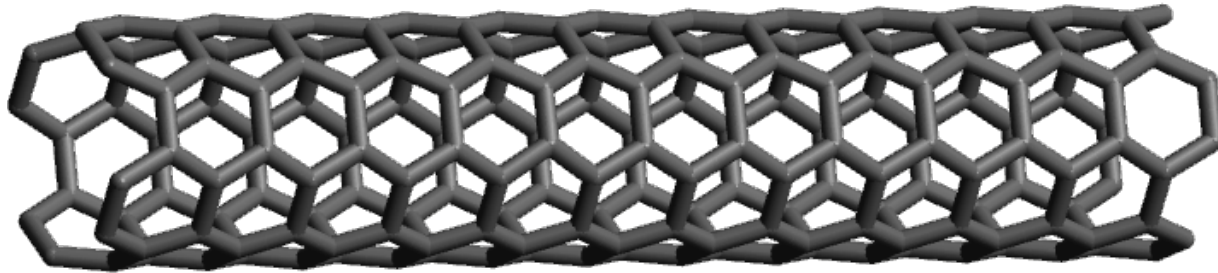
Thermal excitation at room temperature



Molecular dynamics



This effect has been
seen before



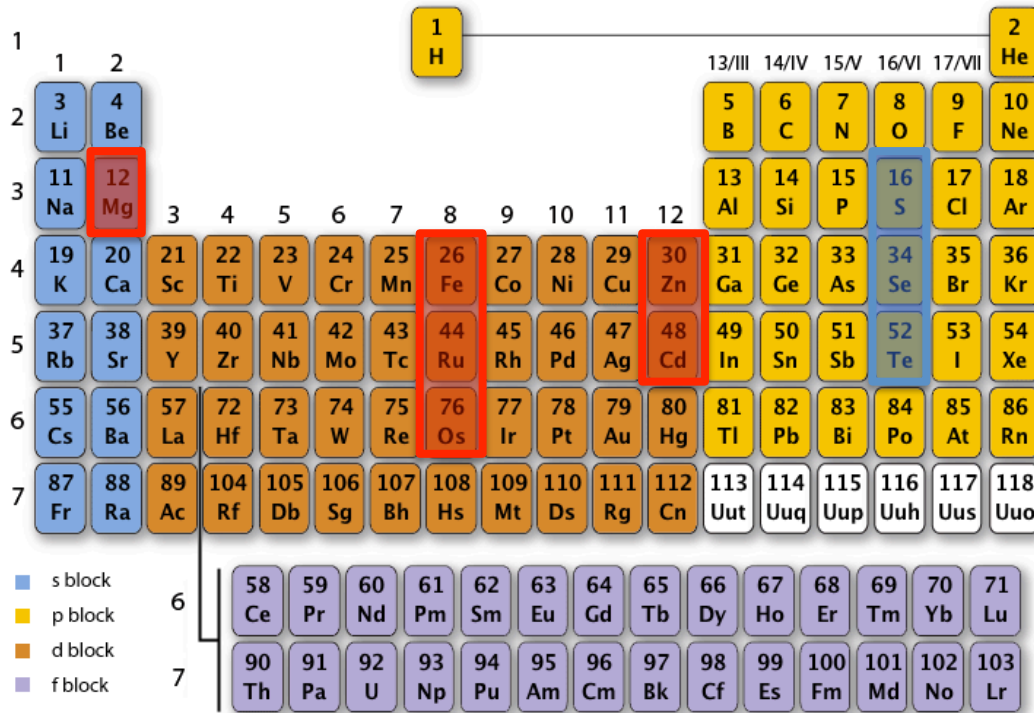
Radial breathing mode

Kim *et al.*, Chem. Phys. **413**, 55 (2013)



Massachusetts Institute of Technology

We predict this effect in other pyrite-structured systems



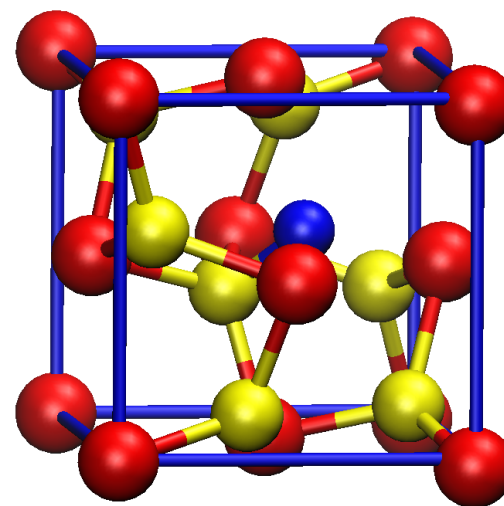
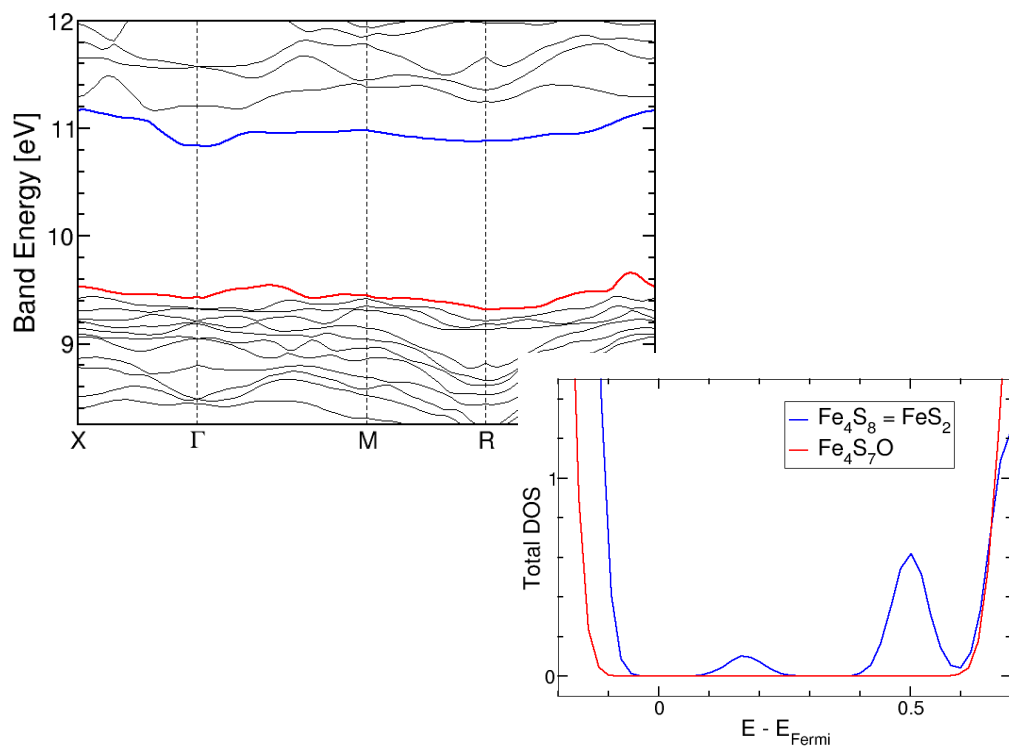
Systems predicted to exhibit oscillating bandgap

FeS ₂	FeSe ₂	FeTe ₂
RuS ₂	RuSe ₂	RuTe ₂
OsS ₂	OsSe ₂	OsTe ₂
ZnS ₂	ZnSe ₂	ZnTe ₂
CdS ₂	MgSe ₂	MgTe ₂



Is there any hope?

Oxygen “defects”



$$E_g^{\text{GW}} = 1.38 \text{ eV}$$



Oxygen defects vs bandgap

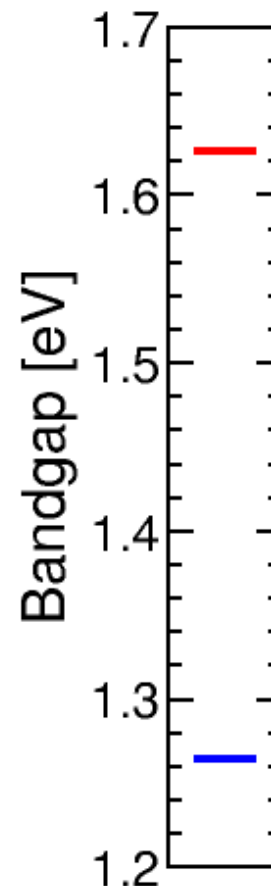
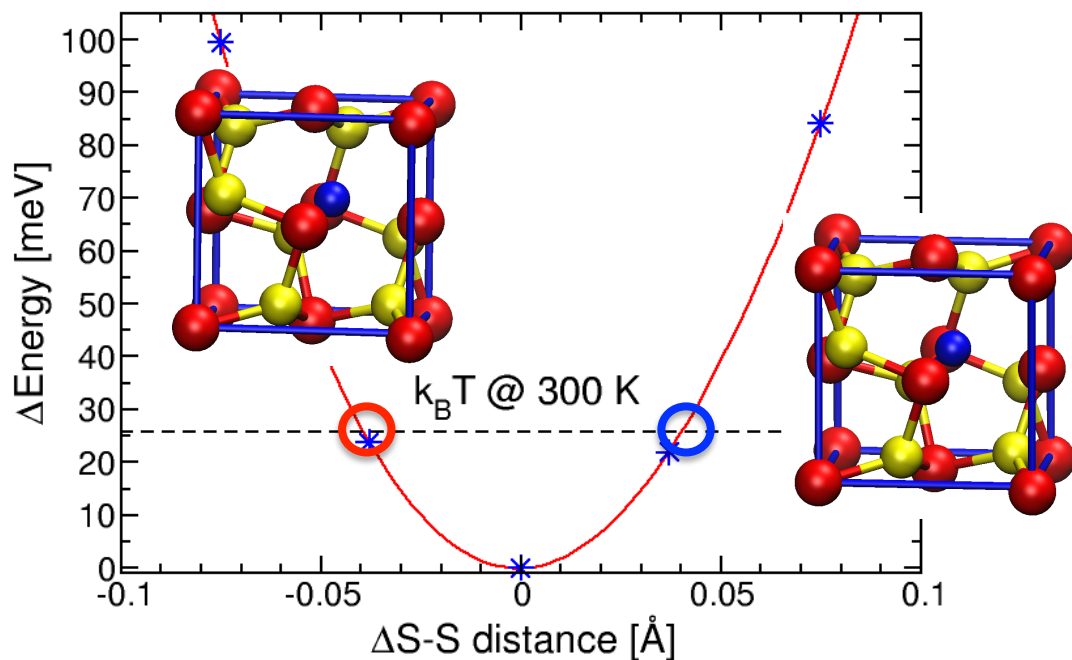
GW calculations

Oxygen/unit cell	Bandgap [eV]
1	1.38
2	1.62
3	1.88
4	2.13

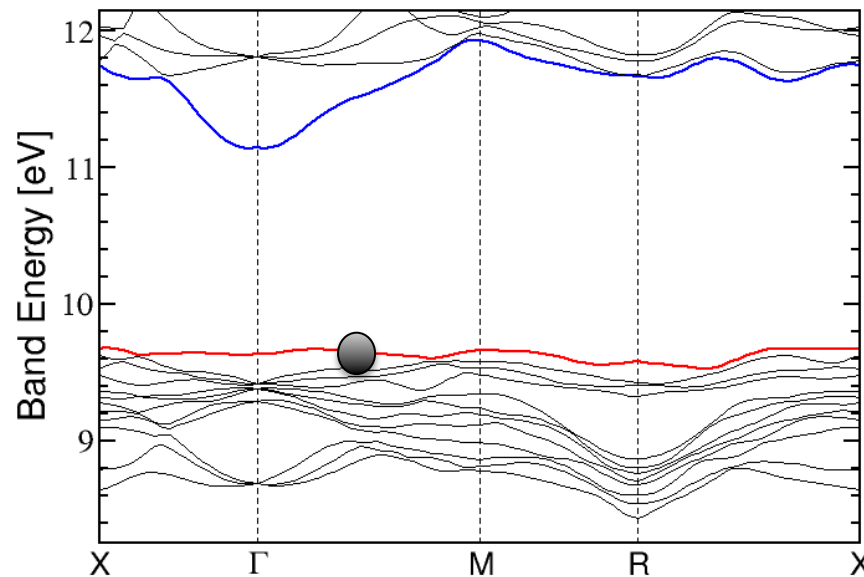


Is there any hope?

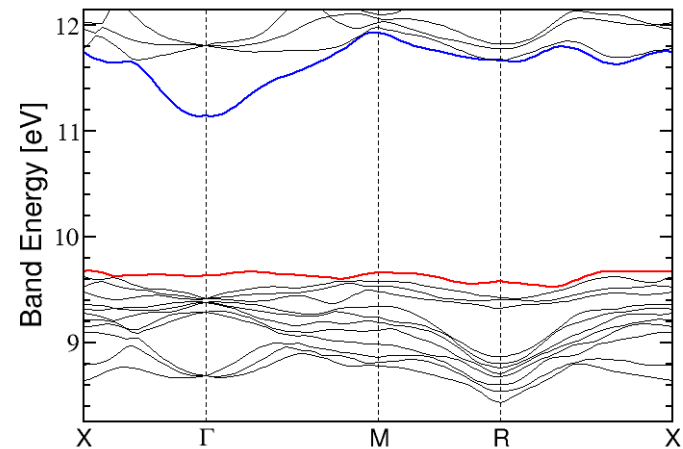
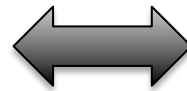
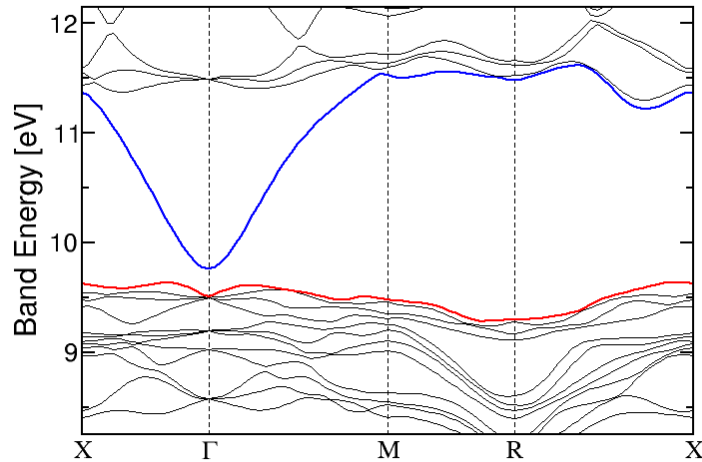
Oxygen "defects"



What about conservation of energy?



What about conservation of energy?



- ~~Thermal population of bands~~
- ~~Energy exchange with phonons~~
- ~~Exchange of electronic kinetic energy~~
- ?



Conclusions

- The bandgap of pyrite depends critically on the sulphur-sulphur distance
- Phonons cause an oscillating bandgap
- Many pyrite-structured systems likely exhibit this oscillation
- Oxygen “defects” cause an opening of the pyrite bandgap and slightly reduce bandgap oscillations



Acknowledgements



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